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Lecture 3:

Energy from fossil fuels

Proven reserves of fossil fuels

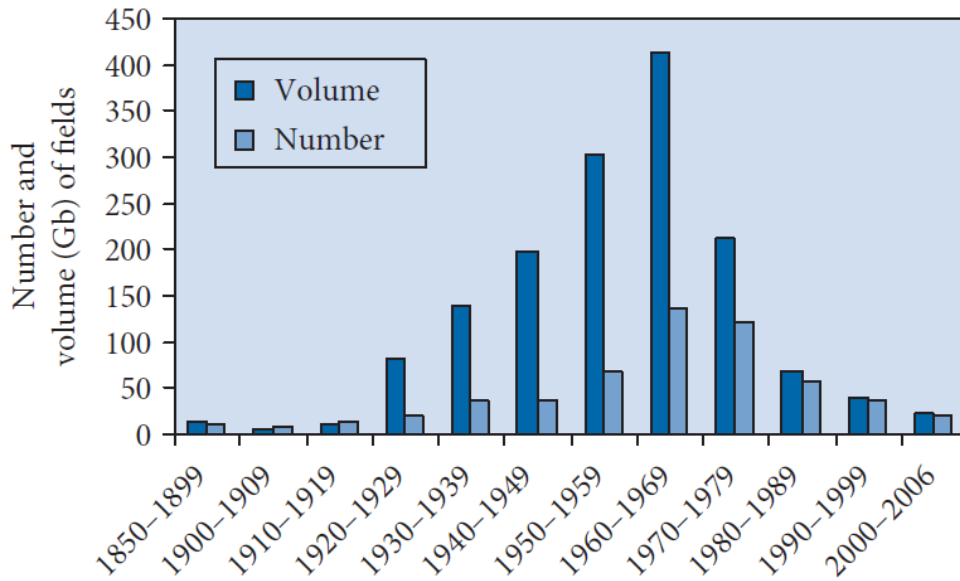
Table 3.1 Proven reserves and R/P values for oil, gas, and coal (239.4 Gt of oil is equivalent to 1697.6 Gb)

Region	Oil (%)	Oil R/P (y)	Gas (%)	Gas R/P (y)	Coal (%)	Coal R/P (y)
North America	14.0	33.1	6.8	13.0	27.5	276
South and Central America	19.4	117.0	4.1	42.5	1.6	150
Europe and Eurasia	9.1	24.4	30.4	57.4	34.8	273
Middle East	47.3	73.1	42.8	129.5	3.7	123
Africa	7.6	42.2	7.5	66.4		
Asia Pacific	2.5	14.0	8.4	28.1	32.3	53
<i>World</i>	239.4 Gt	50.7	186.9 Tm ³	52.8	891.5 Gt	114

Source: BP2016.

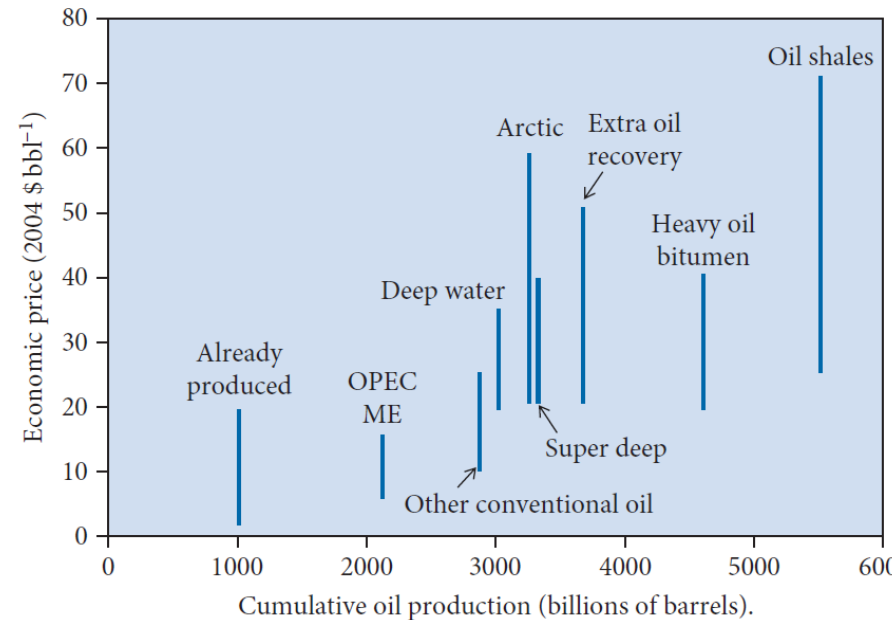
R/P ratio = no. of years that reserves would last if production continued at current level.

Economics of oil and gas production



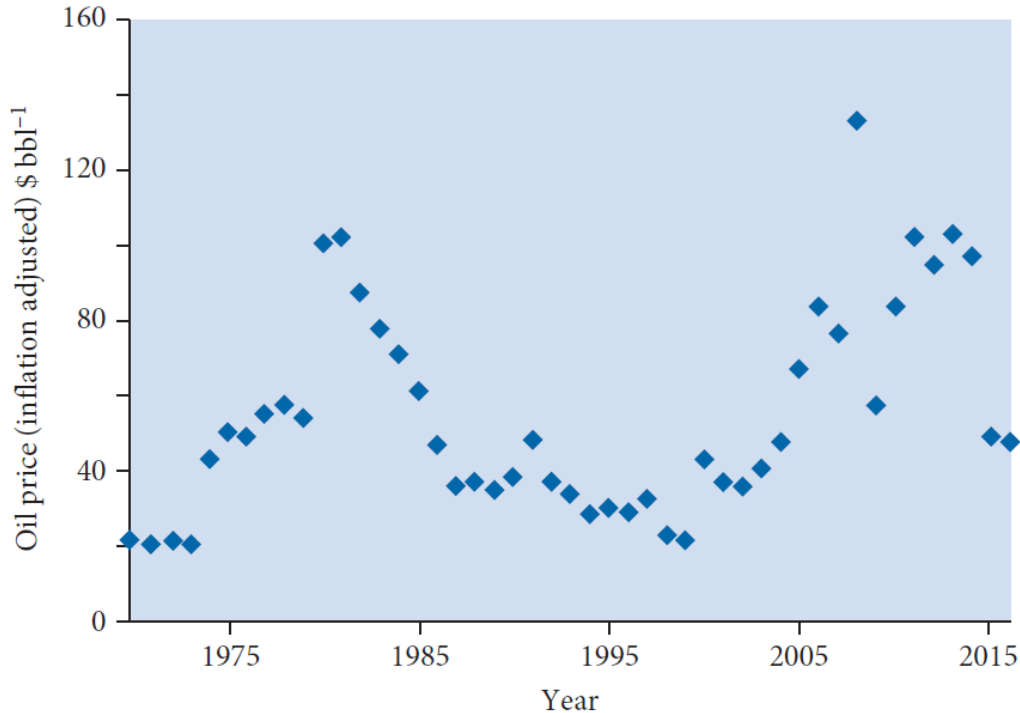
Hubbert peak theory: the production of **conventional** oil and gas rises as new deposits are found and extraction methods improve and falls when deposits become scarcer and more expensive to extract.

In the last 10 years the production of **unconventional** oil and gas (from shale deposits) in USA is reversing this trend.



New deposits are exploited when they can compete with existing deposits

Oil price fluctuations and alternative energy development



Price of oil is very sensitive to political pressures and wars in the Middle East. Periods of high oil prices stimulated R&D into alternative energy sources.

When prices fell back in 90s to previous levels, interest in renewables dropped

Combustion of fossil fuels

Methane (in natural gas): $\text{CH}_4 + 2\text{O}_2 = \text{CO}_2 + 2\text{H}_2\text{O}$

Atomic masses H:O:C = 1:16:12 \Rightarrow 16kg of methane produces 44kg of CO_2

Pentane (in oil): $\text{C}_5\text{H}_{12} + 8\text{O}_2 = 5\text{CO}_2 + 6\text{H}_2\text{O}$

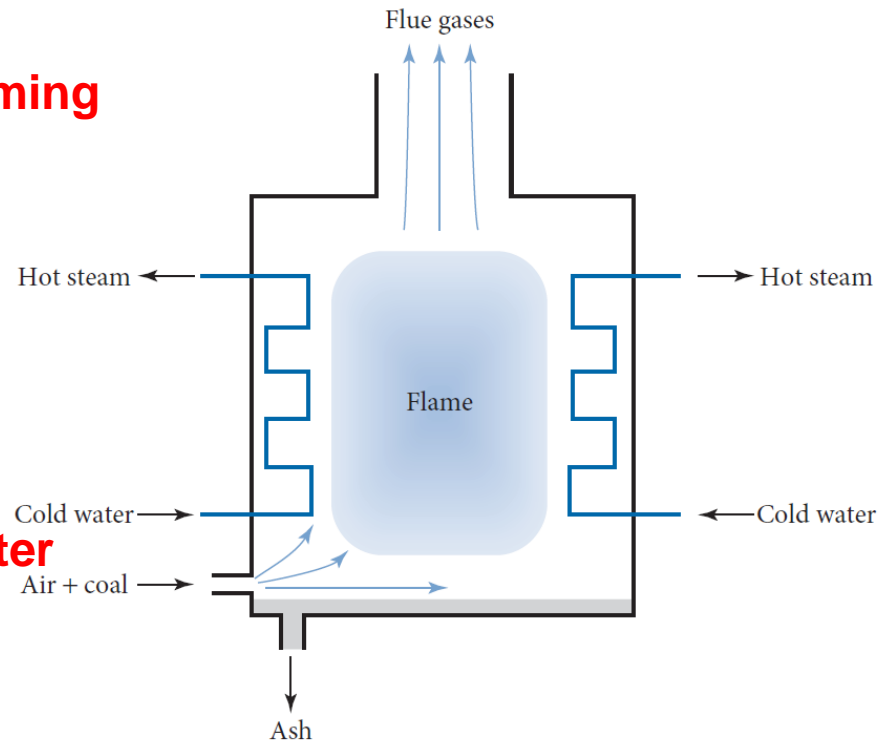
Combustion of fossil fuels releases CO_2 into atmosphere, which contributes to **global warming**

Sulfur in coal produces **acid rain** (damages buildings)

Airborne particulates cause **lung diseases**

Shale oil and shale gas production

- needs vast amounts of **high pressure water** which can cause **minor earthquakes**
- uses **toxic chemicals** which can **pollute groundwater**



Higher and Lower Heating Values

- Heat of combustion = HHV

$$\text{HHV} = \text{LHV} + n \times 44 \text{ kJ mol}^{-1}$$

where n is number of moles water per mole of fuel.

Main contribution is relatively weak double bond in oxygen, rest from stronger O-H compared to C-H

- For compound $\text{C}_c\text{H}_h\text{O}_o$

$$\text{HHV} = -418(c + 0.3h - 0.5o) \text{ kJ mol}^{-1}$$

e.g. octane: $\text{C}_8\text{H}_{18} + 12.5\text{O}_2 \rightarrow 8\text{CO}_2 + 9\text{H}_2\text{O}$

$$\text{HHV} = -418(8 + 18 \times 0.3) = 5601 \text{ kJ mol}^{-1}$$

$$\text{LHV} = \text{HHV} - 9 \times 44 = 5205 \text{ kJ mol}^{-1}$$

Molecular weight 114 g so LHV = 45.7 MJ kg⁻¹ cf 44.4 expt

Carbon capture and storage (CCS)

Large scale carbon capture, e.g. flue gas scrubbing from power plants, presents technical, economic and political challenges. Various long-term options for storage of CO₂ - **carbon sequestration** – are being considered, including

geological: in underground rock formations, e.g. old coal mines and oil fields,
oceans: at depths of more than 3km, where CO₂ liquifies

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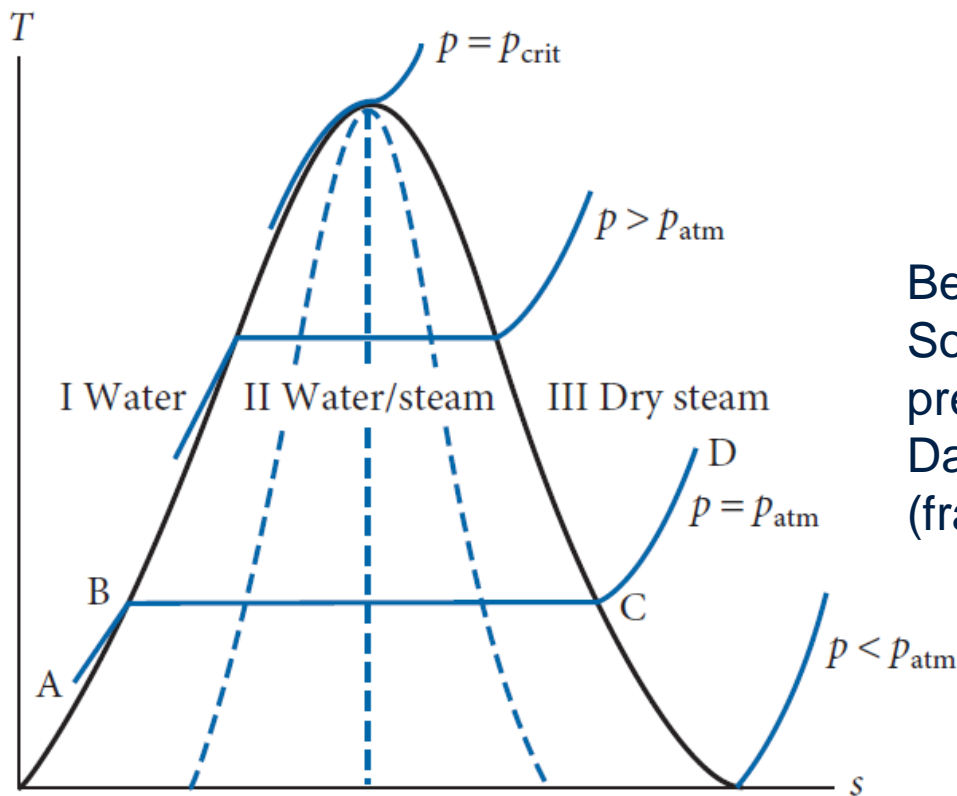
Issues

- CCS technologies are expensive and require international agreements
- CO₂ leakage from underground storage sites
- Sea acidification (carbonic acid)
- Only 8 full-scale CCS projects in 2012, with short timescale to demonstrate technical viability and gain international acceptance

Example: Sleipner CCS project (North Sea) sequesters around 10⁶ tonnes CO₂ per year. Seismic reflection analysis shows distribution of CO₂ is a stable plume trapped between layers of shale.

Thermodynamics of steam power plants

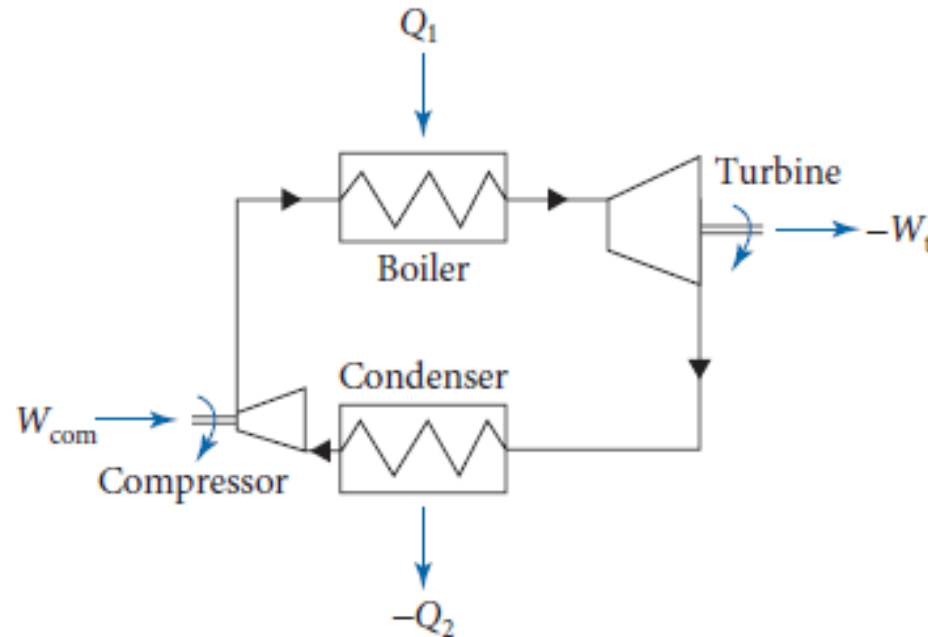
Conventional steam power plants operate in thermodynamic cycle, involving (1) liquid water, (2) 2-phase steam-water mixture, (3) dry steam.



Bell-shaped curve = phase boundary
Solid blue lines = isobars (const. pressure)
Dashed lines = const. steam quality, x (fraction of steam in 2-phase mixture)

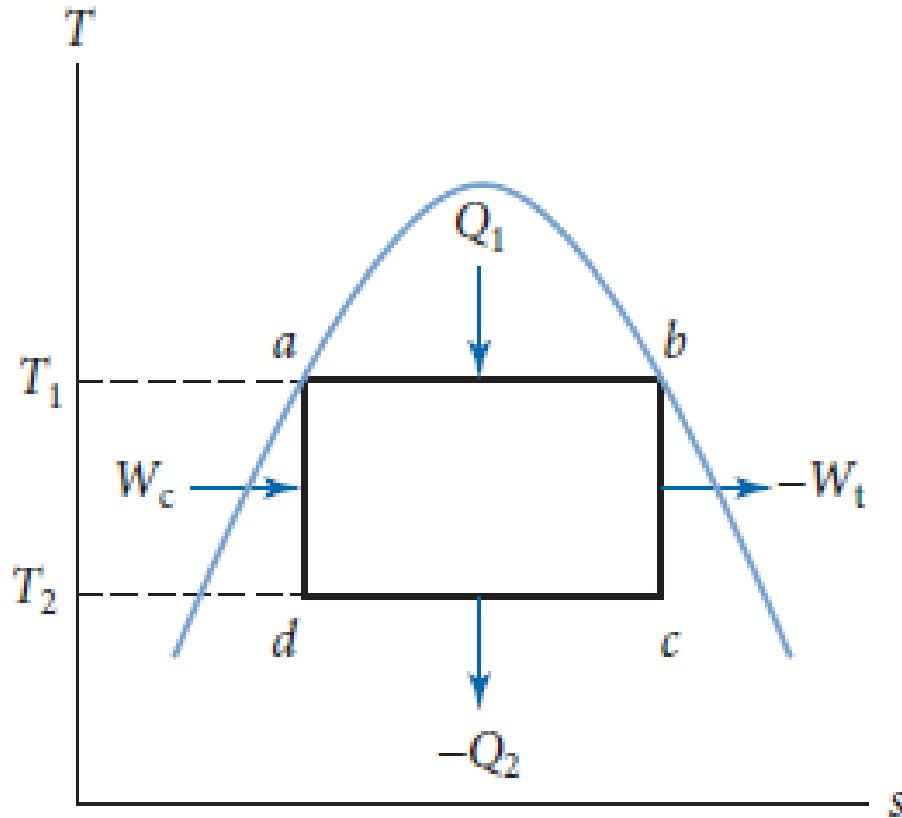
Fig. 3.6 T - s diagram for water and steam (not to scale).

Thermal power plant



- compressor** (also known as the **boiler feed pump**) work W_{com} done on the system to compress cold water from sub-atmospheric pressure to high pressure;
- boiler** heat Q_1 added to the system to convert cold water into steam;
- turbine** work $-W_t$ done by the system (i.e. by steam) on the turbine blades;
- condenser** heat $-Q_2$ lost from the system to the environment in converting steam back to cold water.

Carnot cycle



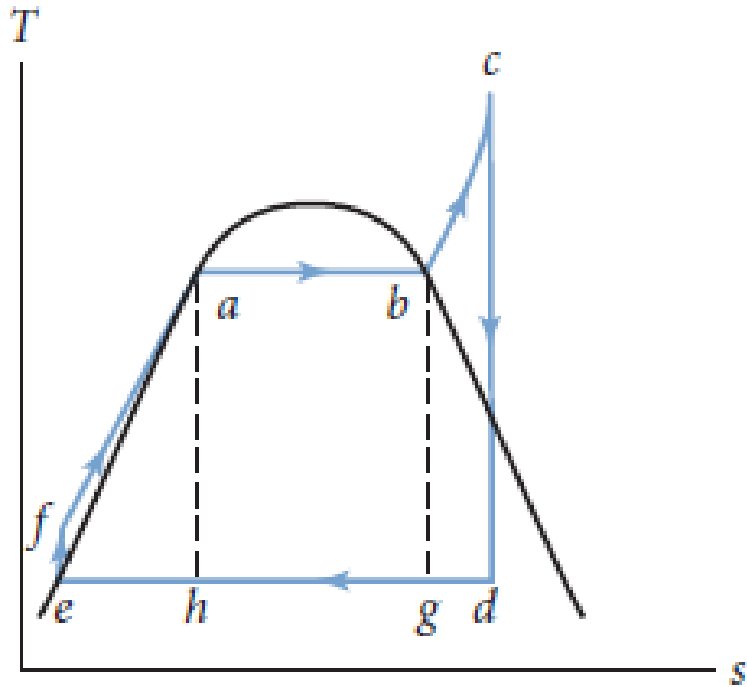
Carnot cycle (constant upper and lower temperatures).

Efficiency is $\eta = 1 - \frac{T_1}{T_2}$

Fixed upper temperature, requires cycle to be in 2-phase region, which lowers efficiency, leads to damage of turbine blades by high momentum water droplets. Also requires enormous compressor.

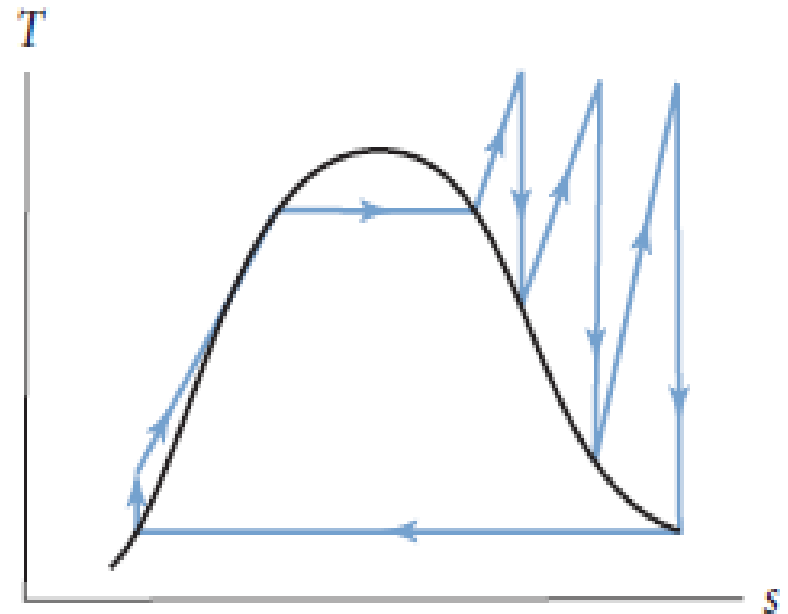
Thermodynamic cycles for steam power plants

Rankine cycle



Rankine cycle without reheat (higher upper temperature than Carnot cycle)

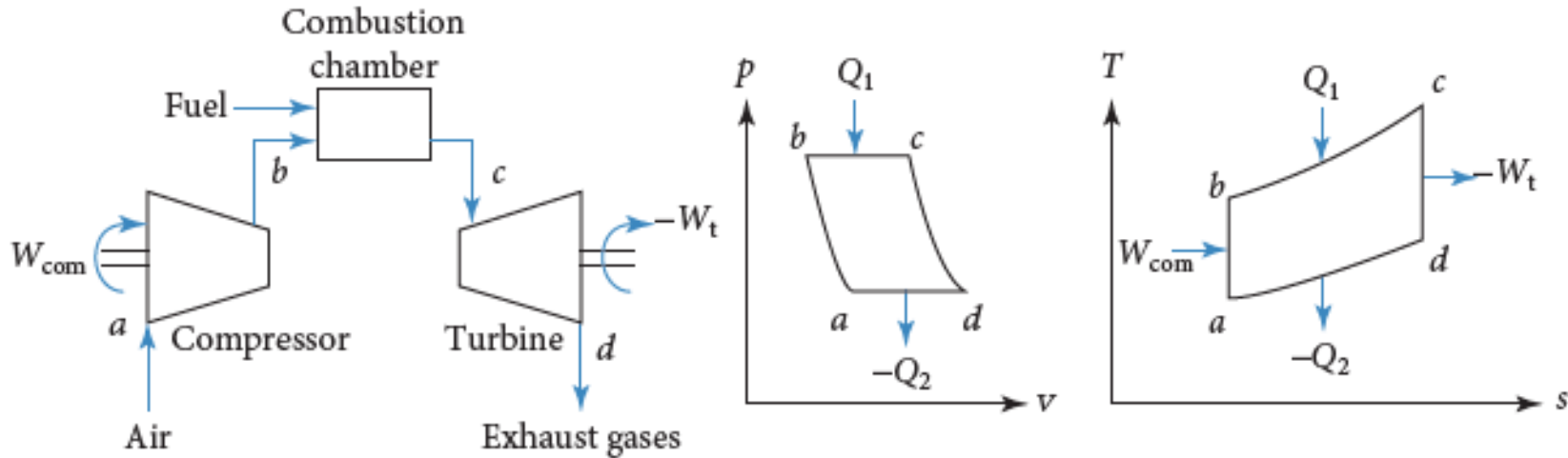
Practical compressor size, but point *d* is inside 2-phase region, so high momentum water droplets not completely eliminated.



Rankine cycle with reheat (3 reheat stages, three turbines: HP, IP, LP)

Avoids water droplets since steam is always dry, upper temperature constrained to around 650° to avoid metallurgical damage, efficiency = 40-45%

Gas turbine plant

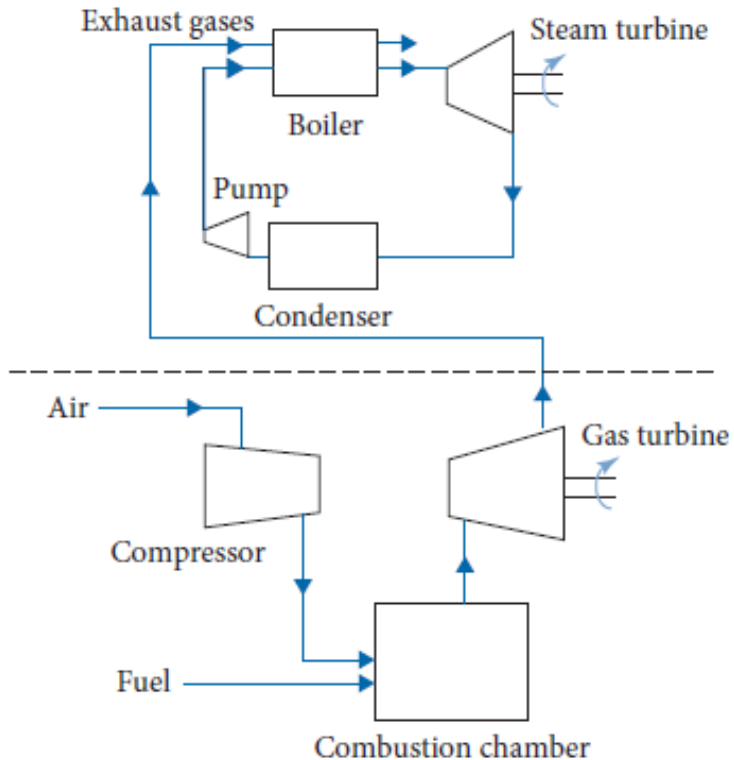


Brayton (or Joule) cycle for gas turbine

Upper end temperature around 1300°C
(turbine blades have ceramic coating and are water-cooled).

Typical efficiency = 45%

Combined cycle gas turbine plant (CCGT)



Combined cycle

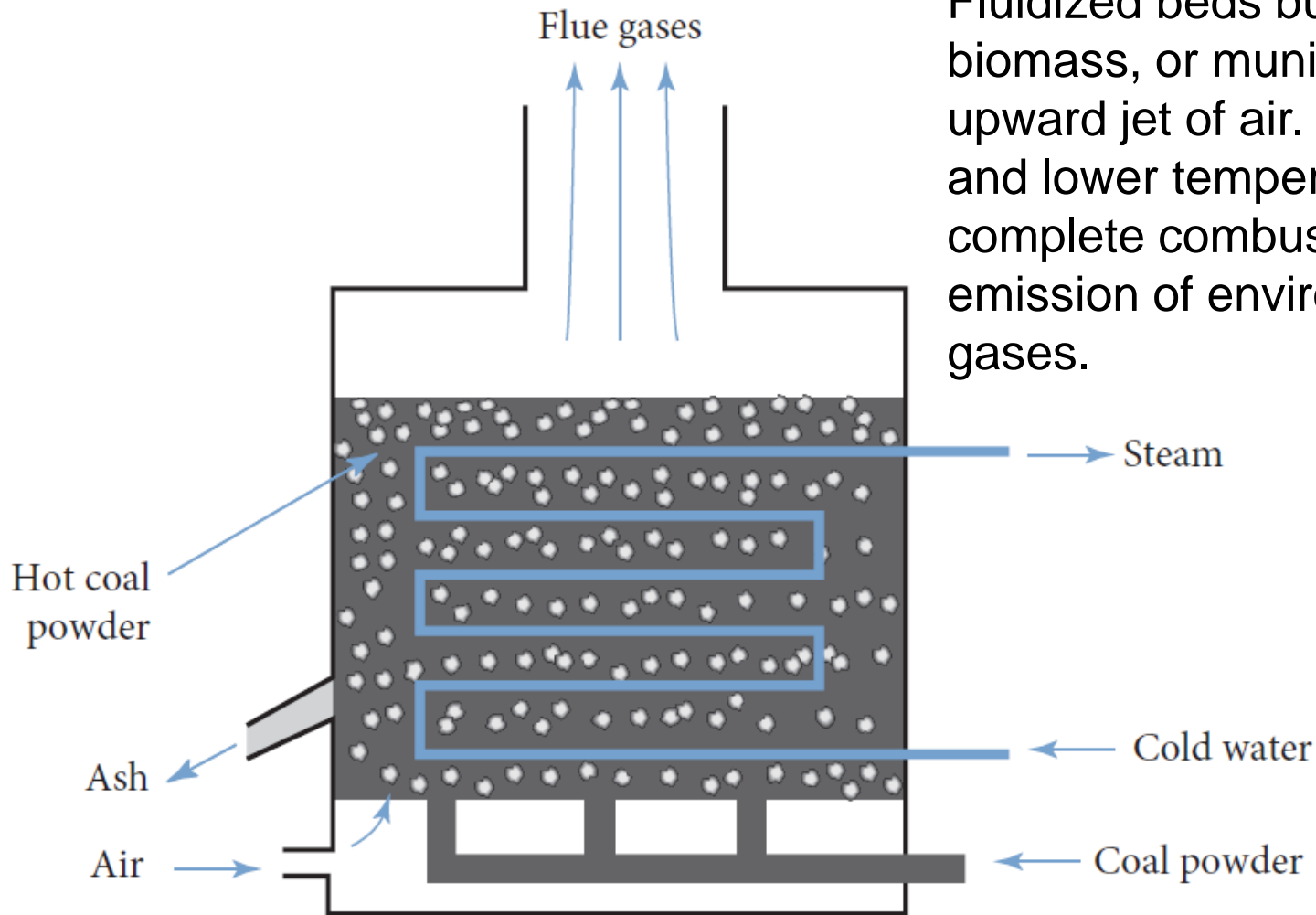


Tranel Kraftwerk Hamm CCGT, Germany

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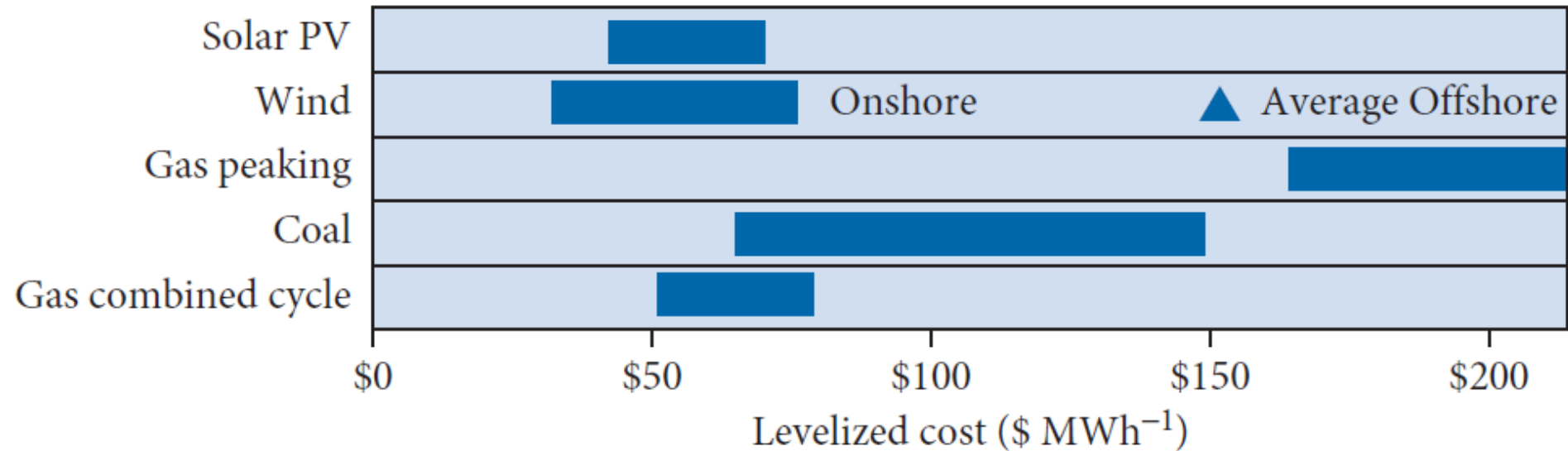
Typical efficiency = 60%, which can be increased to 80% in a combined heat and power cycle (e.g. where waste heat is used for district heating)

Fluidized Beds



Fluidized beds burn pulverized coal, biomass, or municipal waste in upward jet of air. Turbulent mixing and lower temperature more complete combustion reduced emission of environmentally harmful gases.

Economics of fossil fuels



- onshore wind, solar PV are now competitive with fossil-fuel in many locations
- shale oil and gas exploitation has caused shift away from coal in USA
- shale gas will halve CO₂ emissions compared to coal
- India and China account for more than half of the expected 50% of growth in demand up to 2040

Key Points

- The Rankine cycle in power plants overcomes the disadvantages of a Carnot cycle.
- Higher efficiencies can be obtained using combined cycle gas turbines which utilize the waste heat of a Brayton (or Joule) cycle in a Rankine cycle of a steam power plant.
- Fluidized beds can cope with a wide range of feedstock, produce smaller quantities of environmentally harmful gases than conventional combustion chambers, and are now commercially viable.
- It has been estimated that the proven reserves of fossil fuels will last for approximately 51 and 53 years for conventional oil and gas, respectively, and 114 years for coal. The amount of unconventional oil and shale gas approximately doubles the oil and gas reserves, but reserves are determined by limiting emissions rather than resources in the ground.
- Carbon sequestration is a potential means of storing carbon dioxide for long periods, but CCS is expensive and its impact on reducing global warming is uncertain.
- The IEA (2015) estimates that CCS from power generation and in industry could each reduce emissions in 2050 by $\sim 3 \text{ GtCO}_2 \text{ y}^{-1}$, corresponding to $\sim 750 \text{ GWe}$ continuous generation (assuming an equivalence of $\sim 0.5 \text{ kgCO}_2 \text{ kWh}^{-1}$).